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Nucleus Formulas

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List of 11 Nucleus Formulas

Nucleus

1) Average Life

$$\text{fx } t_{avg} = \frac{1}{\lambda}$$

[Open Calculator !\[\]\(a870788d6ed9b8fd294b7654a8c8526b_img.jpg\)](#)

$$\text{ex } 2.5\text{s} = \frac{1}{0.4\text{Hz}}$$

2) Binding Energy

$$\text{fx } BE = (Z \cdot m_p + (A - Z) \cdot m_n - m_{\text{atom}}) \cdot [c]^2$$

[Open Calculator !\[\]\(c50c8b7b2cc2cf9ff925edec0ee94c0d_img.jpg\)](#)

$$\text{ex } 2E^{10}\text{eV} = (17 \cdot 1.00728\text{u} + (37 - 17) \cdot 1.00866\text{u} - 16\text{u}) \cdot [c]^2$$

3) Change in Mass in Nuclear Reaction

$$\text{fx } \Delta m = m_{\text{reactant}} - m$$

[Open Calculator !\[\]\(f60b7a900783ac3fd531bfd9c111be6d_img.jpg\)](#)

$$\text{ex } 3E^{27}\text{u} = 60\text{kg} - 55\text{kg}$$

4) Decay Rate

$$\text{fx } D = -\lambda \cdot N$$

[Open Calculator !\[\]\(83bbbd261710c59db0214aa27b2edc0d_img.jpg\)](#)

$$\text{ex } -26 = -0.4\text{Hz} \cdot 65$$



5) Energy Released in Nuclear Reaction

$$\text{fx } E = \Delta m \cdot [c]^2$$

[Open Calculator !\[\]\(cbe80b694ebd74fcfe136a095b608235_img.jpg\)](#)

$$\text{ex } 1.2E^{-10}\text{J} = 0.8\text{u} \cdot [c]^2$$

6) Half Life for Nuclear Decay

$$\text{fx } t_{0.5} = \frac{0.693}{\lambda}$$

[Open Calculator !\[\]\(3e2231b1ad3ca8da8658228c00dd08e0_img.jpg\)](#)

$$\text{ex } 1.7325\text{s} = \frac{0.693}{0.4\text{Hz}}$$

7) Mass Defect

$$\text{fx } \Delta m = Z \cdot m_p + (A - Z) \cdot m_n - m_{\text{atom}}$$

[Open Calculator !\[\]\(0d5ec72f61334709c3fc9450209b754f_img.jpg\)](#)

$$\text{ex } 21.29696\text{u} = 17 \cdot 1.00728\text{u} + (37 - 17) \cdot 1.00866\text{u} - 16\text{u}$$

8) Nuclear Radius

$$\text{fx } r = r_0 \cdot A^{\frac{1}{3}}$$

[Open Calculator !\[\]\(b64b40baaee5acddc1eab8538ba84754_img.jpg\)](#)

$$\text{ex } 4.165277\text{f} = 1.25\text{f} \cdot (37)^{\frac{1}{3}}$$




9) Population after N Half Lives 

$$\text{fx } N_t = \frac{N_o}{2^n}$$

[Open Calculator !\[\]\(e78f798d4ea5c530c9db49e7d26e6b95_img.jpg\)](#)

$$\text{ex } 1.5625 = \frac{50}{2^5}$$

10) Population at Time t 

$$\text{fx } N_t = N_o \cdot e^{-\frac{\lambda \cdot t}{3.156 \cdot 10^7}}$$

[Open Calculator !\[\]\(05be7c7a8995decd503647c99211f7c2_img.jpg\)](#)

$$\text{ex } 49.99998 = 50 \cdot e^{-\frac{0.4\text{Hz} \cdot 25\text{s}}{3.156 \cdot 10^7}}$$

11) Q-Value 

$$\text{fx } Q = U_i - U_f$$

[Open Calculator !\[\]\(fe3aebe81acea8d45108cd2768939da7_img.jpg\)](#)

$$\text{ex } 10\text{J} = 20\text{J} - 10\text{J}$$



Variables Used






- Δm Mass Defect (*Atomic Mass Unit*)
- **A** Mass Number
- **BE** Binding Energy (*Electron-Volt*)
- **D** Decay Rate
- **E** Energy (*Joule*)
- **m** Mass Product (*Kilogram*)
- m_{atom} Mass of Atom (*Atomic Mass Unit*)
- m_n Mass of Neutron (*Atomic Mass Unit*)
- m_p Mass of Proton (*Atomic Mass Unit*)
- m_{reactant} Mass Reactant (*Kilogram*)
- **n** Number of Half Lives
- **N** Total Number of Particles in Sample
- N_0 Number of Particles in Sample Initially
- N_t Number of Particles at Time t
- **Q** Q Value (*Joule*)
- **r** Nuclear Radius (*Fermi*)
- r_0 Radius of Nucleon (*Fermi*)
- **t** Time (*Second*)
- $t_{0.5}$ Half Life Period (*Second*)
- t_{avg} Average Life (*Second*)
- U_f Final Energy (*Joule*)
- U_i Initial Energy (*Joule*)



- **Z** Atomic Number
- **λ** Decay Constant (Hertz)



Constants, Functions, Measurements used

- **Constant:** [c], 299792458.0 Meter/Second
Light speed in vacuum
- **Constant:** e, 2.71828182845904523536028747135266249
Napier's constant
- **Measurement: Length** in Fermi (f)
Length Unit Conversion 
- **Measurement: Weight** in Atomic Mass Unit (u), Kilogram (kg)
Weight Unit Conversion 
- **Measurement: Time** in Second (s)
Time Unit Conversion 
- **Measurement: Energy** in Electron-Volt (eV), Joule (J)
Energy Unit Conversion 
- **Measurement: Frequency** in Hertz (Hz)
Frequency Unit Conversion 



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